

Enhanced Bimetallic CoM-ZIF/CdS for Effective Photocatalytic Hydrogen Evolution

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Introduction

- **Photocatalyst** is required for the production of renewable hydrogen energy from solar energy since the need for eco-friendly energy is increasing due to global warming.
- **ZIF-67, zeolite imidazole framework using Co metal ions,** has a large surface area, low band gap, high thermal and chemical stability and photocatalytic activity in the visible light region.



CdS is synthesized from the reaction of Cd(NO3)2·6H2O and Na2S·9H2O

The purpose of this study is 1) synthesizing CoM-ZIF/CdS photocatalyst by heterojunction of CdS and transition metals such as Ni, Zn, Cu, and Sn to ZIF-67 and 2) measuring hydrogen production of each photocatalyst by using Gas Chromatography.

Absorbance at wavelength from 300 nm to 800 nm (visible light region)



- CoM-ZIF/CdS(M=Zn, Ni, Cu, Sn) is synthesized by heterojunction of CdS and CoM-ZIF from the 2) reaction of PVP, Co(NO3)2·6H2O, metal nitrate reagent, 2-methylimidazole.
- Hydrogen production is measured by Gas Chromatography and further analysis such as surface 3) observation, visible light absorbance, and band gap is conducted by SEM, Uv-vis spectroscopy.
- Scheme diagram of CoM-ZIF/CdS with by-product, CoS \bullet



Band gap analysis by using Tauc plot



SEM images of ZIF-67, ZIF-67/CdS and CoM-ZIF/CdS (M = Cu, Ni, Zn, Sn)



Average hydrogen evolution rate \bullet



Hydrogen production of each photocatalyst over time



ZIF-67 (a)

(b) ZIF-67/CdS

(c) CoCu-ZIF/CdS

(d) CoNi-ZIF/CdS

CoZn-ZIF/CdS (e)

(f) CoSn-ZIF/CdS

Occonclusions

- ZIF-67/CdS showed better hydrogen production performance than ZIF-67 through heterojunction with CdS.
- CoM-ZIF/CdS showed better hydrogen production performance than ZIF-67/CdS by heterojunction with M = Ni, Zn, Sn. 2)
- CoNi-ZIF/CdS photocatalyst showed the highest hydrogen production because of the high absorbance in visible region with the narrowest band gap. 3)