

Enhanced photoelectrochemical performance from all spin-coated WO₃/BiVO₄/ZnO photoanode for maximizing charge transfer via ZnO decoration control Junhyuk Ji, Yunjin Sung, Harim Kim

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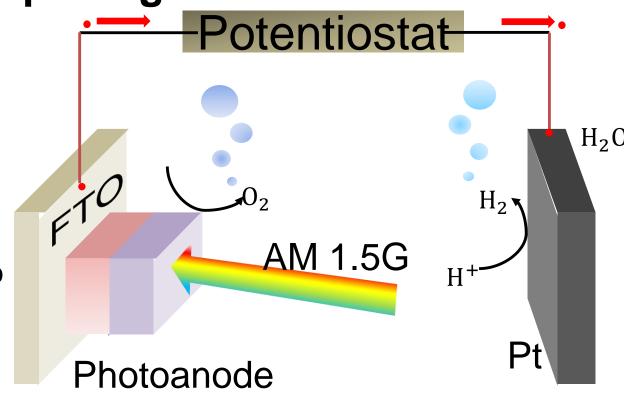
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Introduction

- Photoelectrochemical (PEC) water splitting
- To induce eco-friendly redox reaction by absorbing the solar energy, we can use the solar-activated electrode – photoanode. electrode
- Redox reactions at electrode
- $H_2O + 2h^+ \rightarrow 2H^+ + 0.5O_2$ - Photoanode
- Counter electrode $2H^+ + 2e^- \rightarrow H_2$



- Triple photoanode heterojunction
- To enhance the photoelectrochemical performance, there needs to be proper semiconductor band gap alignment.
- WO₃/BiVO₄ heterojunction has proved to be efficient by many researchers. For improved PEC output, we fabricate the triple photoanode structure by adding the ZnO layer.

Experimental

All materials are prepared by the simple spincoating & annealing process.

- WO₃ - Spin coated Tungstic acid \rightarrow WO₃ by annealing at 500 °C.
- BiVO₄

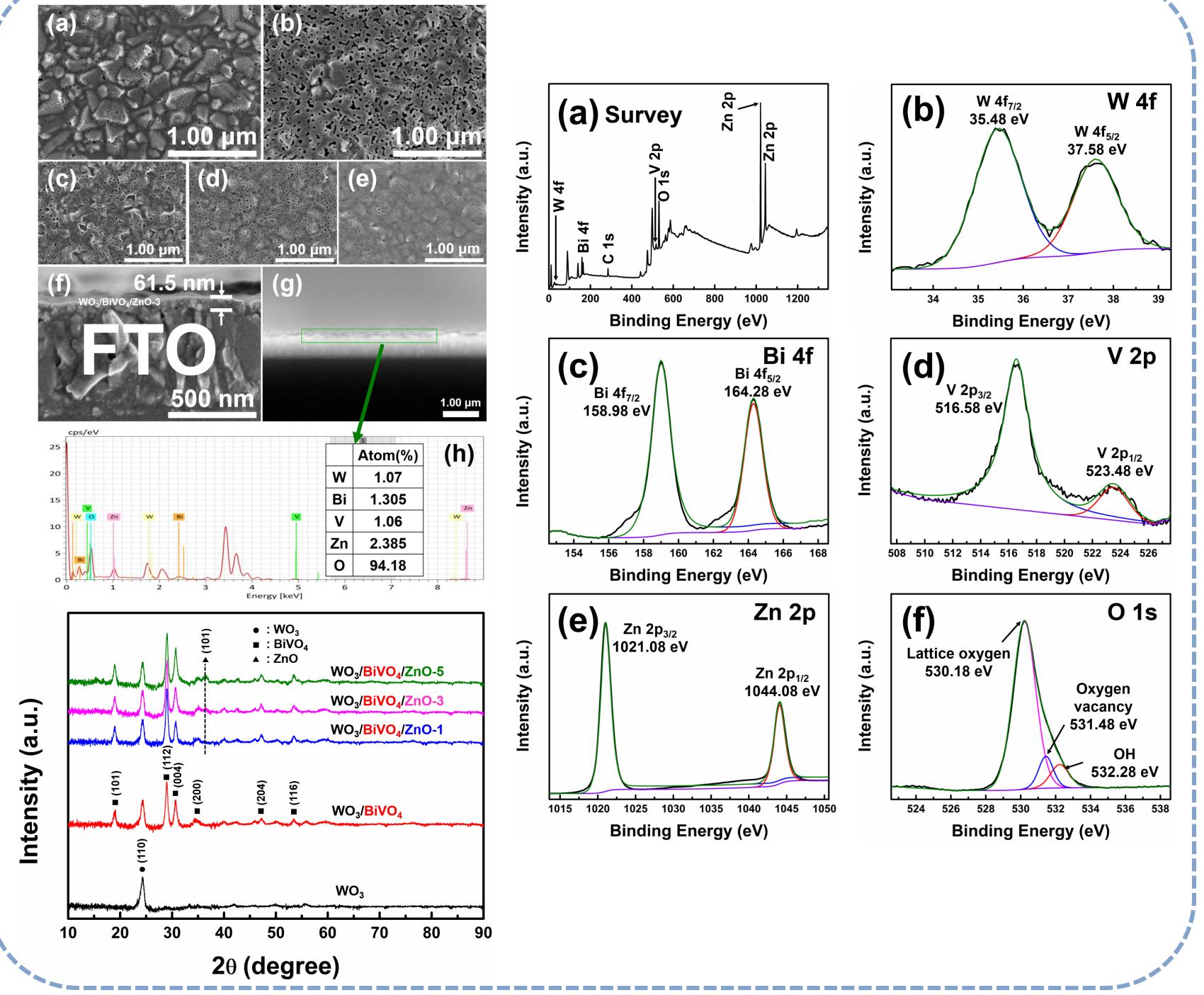
- Spin coated Bismuth nitrate pentahydrate & Ammonium metavanadate \rightarrow BiVO₄ by annealing at 500 °C.

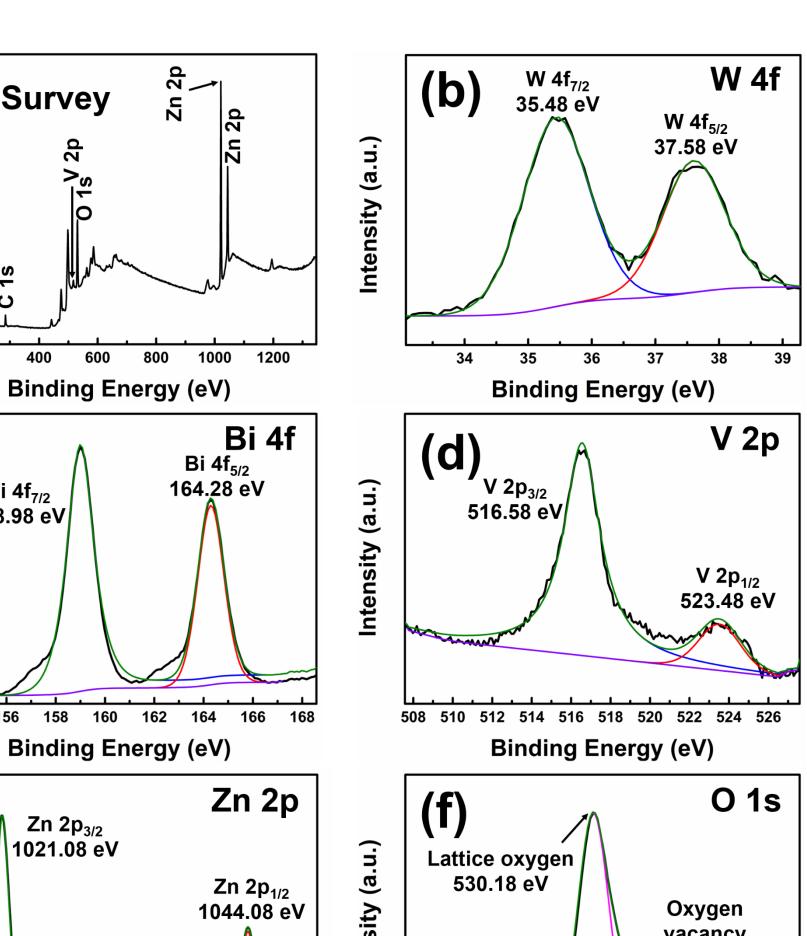
ZnO

- Spin coated Zinc acetate dihydrate \rightarrow ZnO by annealing at 500 °C.

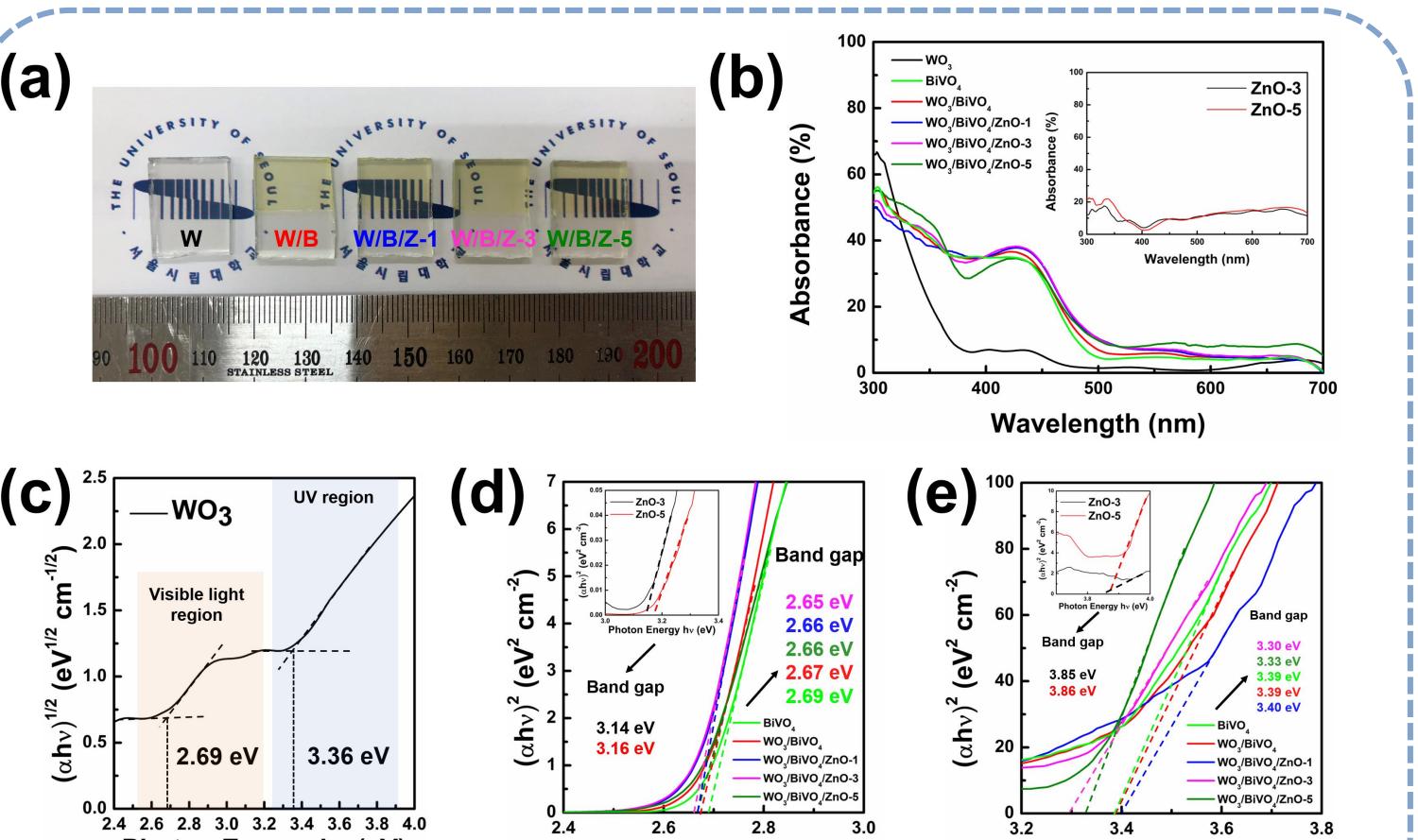
Results & Discussion

Characterizations of photoanodes





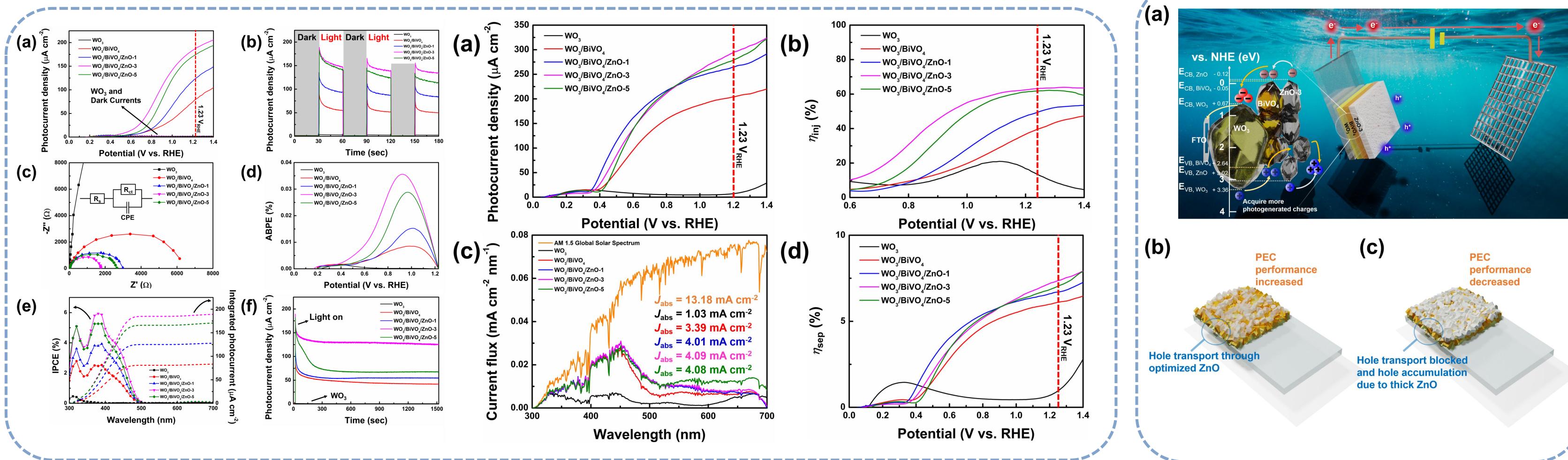
Optical properties



Photon Energy hv (eV) Photon Energy hv (eV) Photon Energy hv (eV)

Photoelectrochemical mechanisms





Conclusion

1) In this study, we investigate the effects of the number of the ZnO spin-coatings.

2) PEC performance of the WO₃/BiVO₄ type II photoanode can be improved by applying optimized ZnO components with proper band gap alignment.

3) This result implies that the ZnO layer can suppress the charge recombination and improve the charge separation efficiency. Furthermore, it can passivate the photoanode surface against the photocorrosion.

Reference

- 1. N. Liu, M. Han, Y. Sun, C. Zhu, Y. Zhou, Y. Zhang, H. Huang, V. Kremnican, Y. Liu, Y. Lifshitz and Z. Kang, *Energy Environ. Sci.*, 2018, **11**, 1841–1847. 2. K. K. Dey, S. Gahlawat and P. P. Ingole, J. Mater. Chem. A, 2019, 7, 21207-21221.
- 3. S. Chatterjee, P. Bhanja, D. Ghosh, P. Kumar, S. Kanti Das, S. Dalapati and A. Bhaumik, *ChemSusChem*, 2020, 1–10.

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