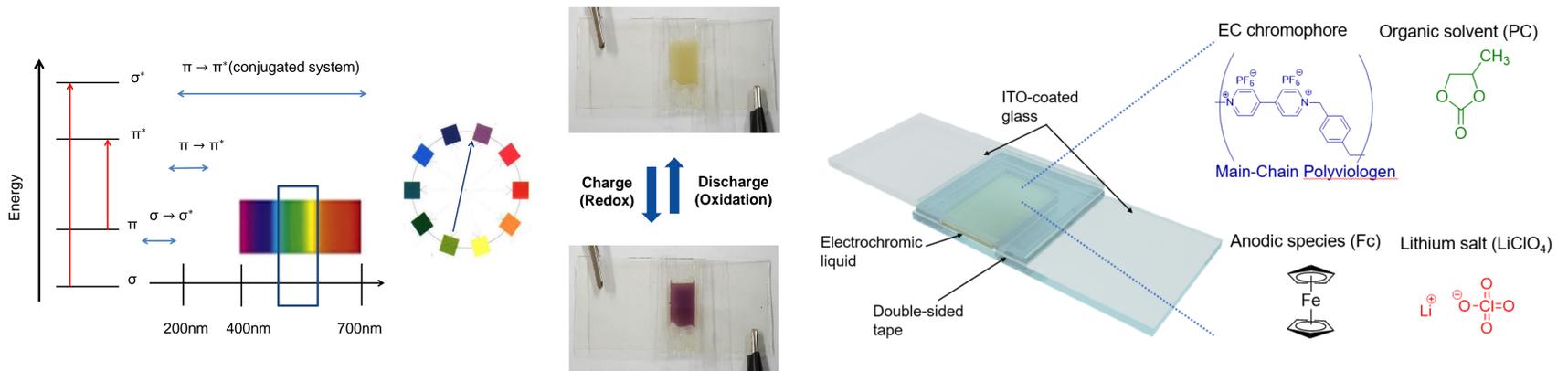


Introduction & Objective

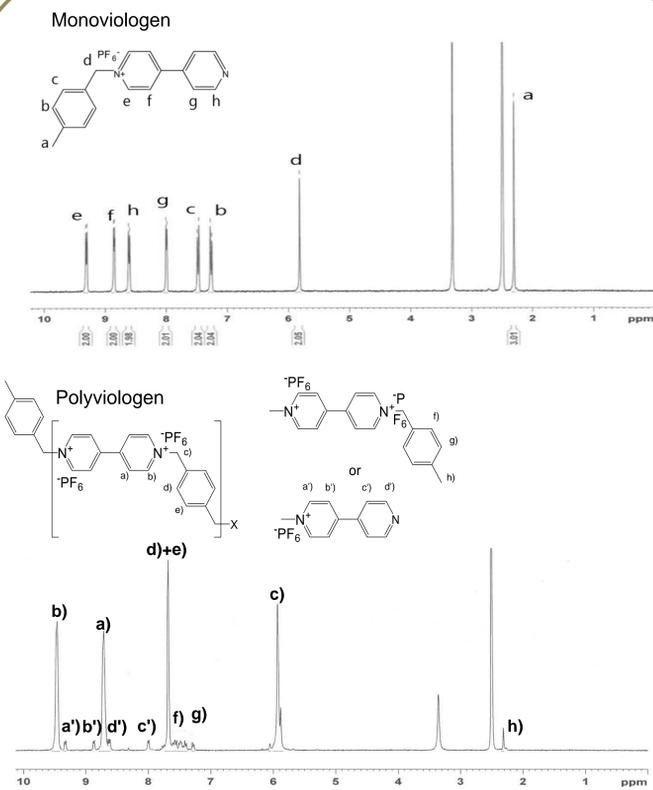
Objective

- Find polyviologen which has optimum performance for supercapacitor.
- Produce simple supercapacitor using polyviologen.
- Check its performance and compare with comparative group.
- Check its coloration characteristic.

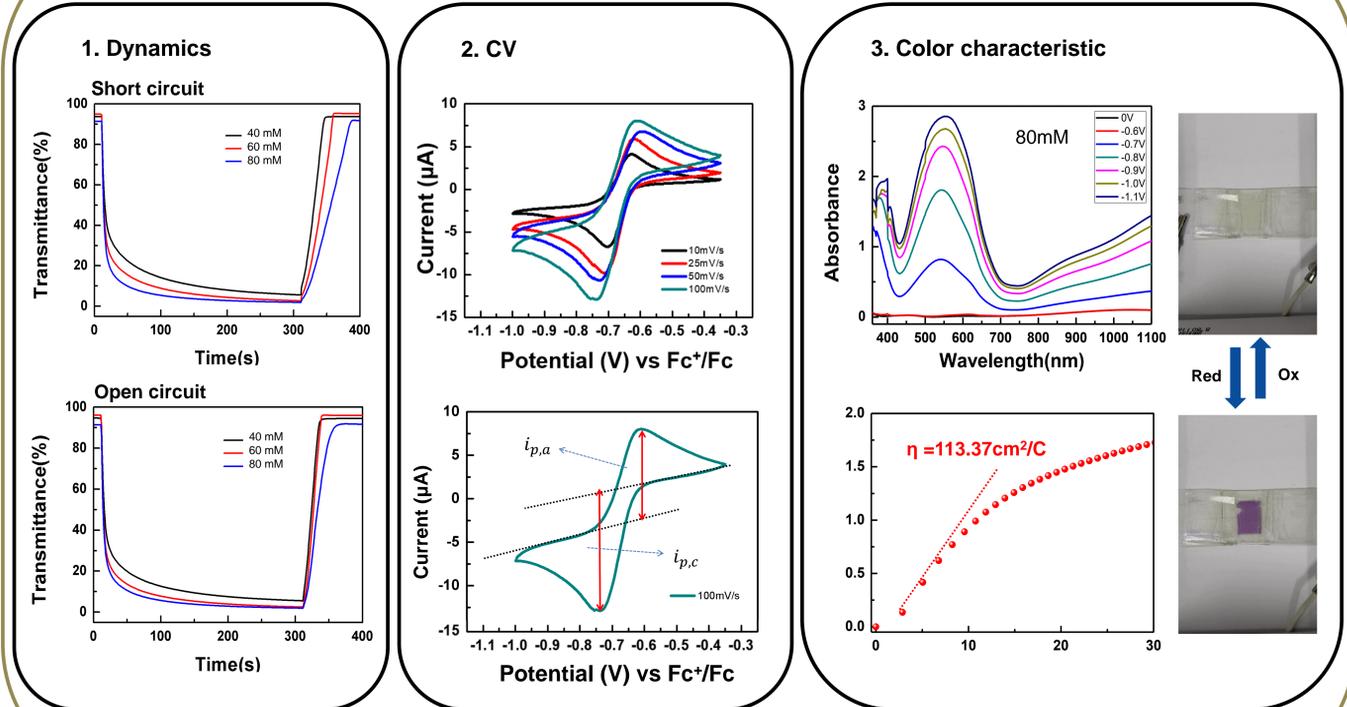


Results and Discussion

NMR analysis



Performance of polyviologen



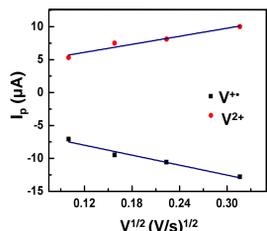
Diffusion Coefficient

Randles-Sevcik Eq.

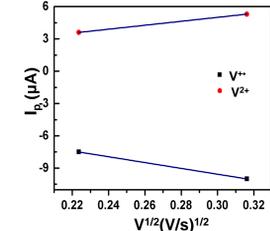
$$i_p = 0.4463 \cdot n \cdot F \cdot A \cdot C \cdot \left(\frac{nFvD}{RT} \right)^{0.5}$$

i_p = current maximum in amps
 n = number of electrons transferred in the redox event (usually 1)
 A = electrode area in cm^2
 F = Faraday Constant in C mol^{-1}
 D = diffusion coefficient in cm^2/s
 C = concentration in mol/cm^3
 v = scan rate in V/s
 R = Gas constant in $\text{J K}^{-1} \text{mol}^{-1}$
 T = temperature in K

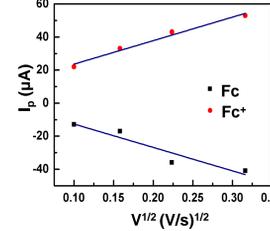
1. Polyviologen



2. Monoviologen

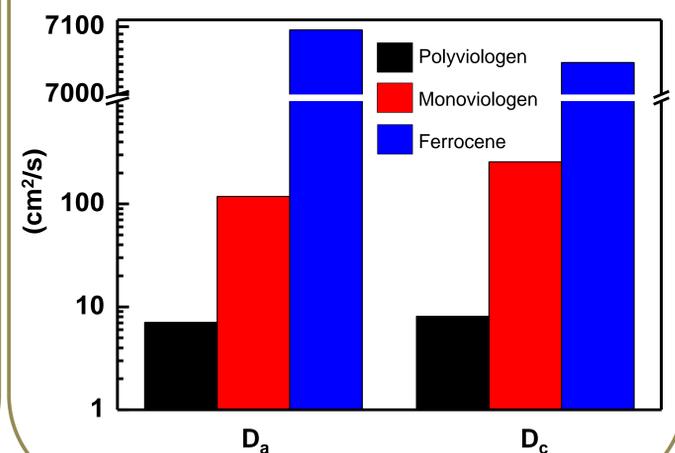


3. Ferrocene



Conclusion

Diffusion coefficient



Summary

- We proposed a simple yet highly effective supercapacitor using polyviologen.
- Electrochemical and electrochromic properties of polyviologen were verified using dynamics, cyclic voltammetry and UV/vis spectroscopy.
- Polyviologen was suitable for supercapacitor due to smaller diffusion coefficient than that of monoviologen.
- In order to improve the performance of supercapacitor, an anodic species with a small diffusion coefficient should be developed.